

### Preliminary communication

## SYNTHESIS OF CARBORANYL DERIVATIVES OF DI- AND TRI-VALENT LANTHANIDES

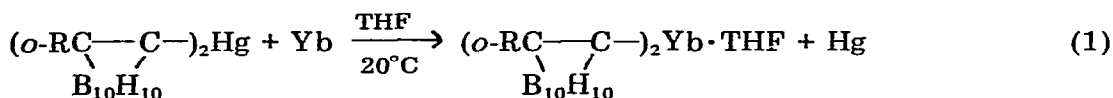
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### Summary

It has been shown that the interaction of *C*- and *B*-mercurocarboranes with lanthanides (La, Tm, Yb) in THF at 20°C give carboranyl derivatives of lanthanides in which the lanthanides are linked with the carborane nuclei via C—Ln or B—Ln bonds.

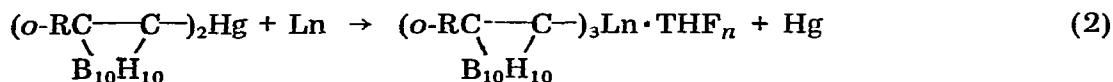
In continuation of investigations on the synthesis of new derivatives of lanthanides [1] we have studied the reaction of *C*- and *B*-mercurocarboranes with Ln<sup>0</sup>, where Ln = La, Tm, Yb.

It is known that ytterbium(0) reacts with R<sub>2</sub>Hg to give stable organic derivatives in the divalent state [2,3]. Investigations of the reaction of symmetrical *C*-mercuro derivatives of methyl- and phenylcarboranes with ytterbium(0) showed that these compounds react in THF at 20°C to give *C*-carboranyl derivatives of ytterbium(II) in 30–50% yield (eq. 1).



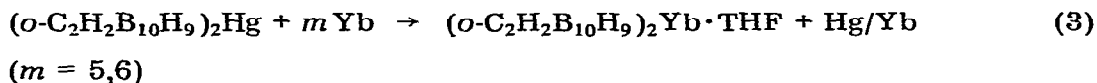
(R = CH<sub>3</sub>, C<sub>6</sub>H<sub>5</sub>)

Under similar reaction conditions *C*-mercuro derivatives of carboranes react with lanthanum(0) or thulium(0) to form carboranyl derivatives of trivalent lanthanum or thulium (eq. 2).



(R = CH<sub>3</sub>, C<sub>6</sub>H<sub>5</sub>; Ln = La, Tm; n = 1–3)

*B*-Mercurio derivatives of carboranes undergo the same reaction to give in the case of Yb carboranyl derivatives of ytterbium(II) linked with the carborane nucleus via a boron atom. However, this reaction takes place with large excess (5–6 times) of activated Yb only (eq. 3)



With smaller quantities of Yb the reaction proceeds very slowly and leads to a mixture of products which are difficult to separate. We found also that unsymmetrical *B*-mercuriocarboranyl derivatives react with thulium to give carboranyl derivatives of thulium with B–Tm bond (eq. 4), but it is also possible that the formation of B–Tm–C derivative is followed by disproportionation.

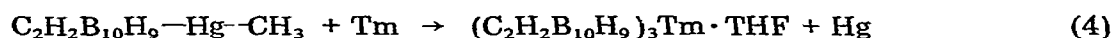


Table 1 contains the physical data for the compounds obtained.

TABLE 1  
CARBORANYL DERIVATIVES OF LANTHANIDES

Compounds <sup>a</sup>	Decomposition point (°C)	Yield (%)	Analysis (Found (calcd.) (%))	
			C	H
(C <sub>6</sub> H <sub>5</sub> CB <sub>10</sub> H <sub>10</sub> C) <sub>3</sub> La·THF	260	32	38.7 (38.7)	7.4 (6.1)
(C <sub>6</sub> H <sub>5</sub> CB <sub>10</sub> H <sub>10</sub> C) <sub>3</sub> Tm·3THF	275	39	42.6 (41.8)	5.2 (6.8)
(CH <sub>3</sub> CB <sub>10</sub> H <sub>10</sub> C) <sub>3</sub> Tm·3THF	280	45	28.3 (29.3)	7.8 (7.4)
(C <sub>6</sub> H <sub>5</sub> CB <sub>10</sub> H <sub>10</sub> C) <sub>2</sub> Yb·2THF	262	29	34.7 (33.8)	5.6 (5.4)
(CH <sub>3</sub> CB <sub>10</sub> H <sub>10</sub> C) <sub>2</sub> Yb·2THF	287	47	22.0 (22.9)	4.7 (5.7)
(C <sub>2</sub> H <sub>2</sub> B <sub>10</sub> H <sub>9</sub> ) <sub>3</sub> Tm·THF	265	34	18.8 (17.9)	5.0 (5.9)
(C <sub>2</sub> H <sub>2</sub> B <sub>10</sub> H <sub>9</sub> ) <sub>2</sub> Yb·THF	290	22	17.7 (18.4)	6.0 (5.7)

<sup>a</sup>The *ortho*-isomer of carborane was used.

## References

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- 2 G.B. Deacon, A.J. Koplick, W.D. Raverty and D.G. Vince, *J. Organometal. Chem.*, 182 (1979) 121.
- 3 G.Z. Suleimanov, T.Kh. Kurbanov, Ya.A. Nuriev, L.F. Rybakova and I.P. Beletskaya, *Dokl. Akad. Nauk, SSSR*, in press.